



## Influence of terrestrial weathering on ocean acidification and the next glacial inception

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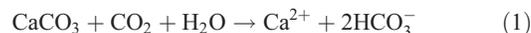
[1] Ocean uptake of anthropogenic CO<sub>2</sub> causes a decline in seawater pH, a process known as ocean acidification, which may adversely affect marine organisms. We investigate whether continental weathering can mitigate future ocean acidification by sequestering atmospheric CO<sub>2</sub>. We conducted simulations under a suite of carbon emission scenarios with different weathering parameterizations. The short-term impact of a strong weathering feedback was only notable for large emissions with slow injection. This mitigation by enhanced weathering, however, is an order of magnitude smaller than the expected maximum pH decline based on the default parameterizations. Thus on short timescales, weathering has little effect on future atmospheric CO<sub>2</sub> and ocean acidification, regardless of the assumed weathering feedback strength. But on longer timescales and for large emissions, different weathering parameterizations introduce large uncertainties regarding the time when pCO<sub>2</sub> will return to climatically relevant levels of, say, 400 μatm in the future. **Citation:** Uchikawa, J., and R. E. Zeebe (2008), Influence of terrestrial weathering on ocean acidification and the next glacial inception, *Geophys. Res. Lett.*, 35, L23608, doi:10.1029/2008GL035963.

### 1. Introduction

[2] Humans have released over 300 Pg of carbon (1 Pg C = 10<sup>15</sup> g C) mainly from fossil fuel combustions [Marland *et al.*, 2007], and the emission is still ongoing at accelerating rates [Archer, 2005; Intergovernmental Panel on Climate Change (IPCC), 2007]. Much of the anthropogenic CO<sub>2</sub> is absorbed by the ocean, which alleviates rapid climate changes associated with global warming. But this buffering by the ocean unfortunately has its price: ocean acidification. Uptake of atmospheric CO<sub>2</sub> lowers seawater pH and [CO<sub>3</sub><sup>2-</sup>], and the saturation state of CaCO<sub>3</sub> [Caldeira and Wickett, 2003; Orr *et al.*, 2005; Zeebe *et al.*, 2008]. After absorbing ~40% of anthropogenic CO<sub>2</sub> over the last 200 years, average surface ocean pH has dropped by 0.1 units [IPCC, 2007; Zeebe *et al.*, 2008]. If the world fossil fuel reserves are fully exploited in the future (~5,000 Pg C of emission [Sundquist, 1986; Rogner, 1997]), surface ocean pH can drop by ~0.6 to ~0.78 units [Zeebe *et al.*, 2008]. Further ocean acidification may particularly threaten marine calcifying organisms [i.e., Riebesell *et al.*, 2000; Orr *et al.*, 2005; Hoegh-Guldberg *et al.*, 2007].

[3] But a future increase in pCO<sub>2</sub> and consequential warming should enhance carbonate and silicate weathering

[Walker *et al.*, 1981; Berner *et al.*, 1983; White and Blum, 1995], thus removing atmospheric CO<sub>2</sub>, followed by CaCO<sub>3</sub> burial in sediments:



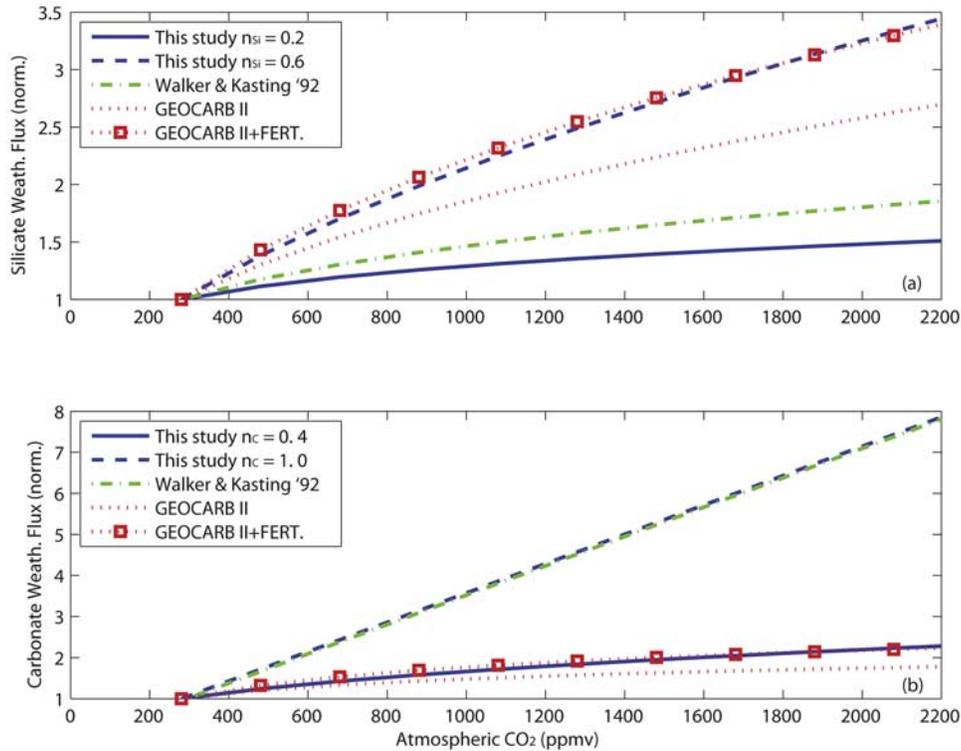
This negative feedback involving weathering of continental materials played a dominant role in regulating pCO<sub>2</sub> on geologic timescales [Walker *et al.*, 1981; Berner *et al.*, 1983; Zachos *et al.*, 2008] and it is likely to remain so in response to future anthropogenic C emissions [Archer, 2005; Lenton and Britton, 2006]. Then, would this weathering feedback be effective in alleviating future ocean acidification by buffering the decline in ocean pH? This question is largely unresolved because weathering fluxes were typically prescribed as constant [i.e., Archer *et al.*, 1998; Archer, 2005; Ridgwell and Edwards, 2007] or omitted in previous model simulations [i.e., Caldeira and Wickett, 2005; Montenegro *et al.*, 2007; Tyrrell *et al.*, 2007]. Studies by Walker and Kasting [1992], Lenton and Britton [2006] and Zeebe *et al.* [2008] represent a few model simulations in which weathering fluxes were set to co-vary with pCO<sub>2</sub> (and with plant productivity in Lenton and Britton [2006]). Walker and Kasting [1992] and Lenton and Britton [2006] largely focused on the impact of the weathering feedback on pCO<sub>2</sub>; however, less attention was given to ocean acidification. Zeebe *et al.* [2008] modeled future ocean acidification but with a single set of parameterizations to relate carbonate and silicate weathering fluxes to pCO<sub>2</sub>.

[4] To determine whether enhanced weathering feedback can mitigate the future rise in pCO<sub>2</sub> and ocean acidification, we conducted simulations using the Long-term Ocean-Atmosphere-Sediment Carbon cycle Reservoir (LOSCAR) model [Zeebe *et al.*, 2008; Zachos *et al.*, 2008] under various weathering parameterizations. We also conducted 300,000-year long simulations to test the effect of weathering parameters on predictions of the long-term return to climatically relevant atmospheric CO<sub>2</sub> levels. For example, Lunt *et al.* [2008] suggest that the growth of Northern Hemisphere ice sheets can be triggered if atmospheric CO<sub>2</sub> levels fall below a threshold of ~400 ppmv. Thus, this value could have important implications for the timing of the next glacial inception.

### 2. Model Descriptions and Weathering Parameterizations

[5] The LOSCAR model is a carbon-cycle reservoir model (modified from Walker and Kasting [1992]) coupled

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**Figure 1.** Comparison of (a) silicate and (b) carbonate weathering fluxes as a function of atmospheric CO<sub>2</sub> used in the LOSCAR model (this study), the GEOCARB II and GEOCARB II+FERT model [Berner, 2001] and the Walker and Kasting [1992] model. The weathering fluxes shown in the plots are normalized to initial weathering fluxes of 1.

to a sediment module [Zeebe and Zachos, 2007]. The model runs were performed from year 1,700 (pre-industrial condition) to 10,000 (or 300,000) under a suite of C emission scenarios based on historic total emissions of 315 Pg C until year 2004 [Marland et al., 2007] and projected future emissions ranging from 600 to 5,000 Gt C total emissions released over 200 to 1,000 years. Simulations over the next few centuries by the LOSCAR model [Zeebe et al., 2008; Zachos et al., 2008] are in good agreement with the results of previous carbon-cycle model studies [Caldeira and Wickett, 2003; Orr et al., 2005; Montenegro et al., 2007]. However, our estimates of atmospheric CO<sub>2</sub> rise and seawater pH decline for a given emission scenario are more conservative because the model incorporates dissolution of sedimentary CaCO<sub>3</sub> and a weathering feedback.

[6] In the model, silicate and carbonate weathering fluxes ( $F_{Si}$  and  $F_C$ ) are expressed as:

$$F_{Si} = F_{Si}^0 \times (pCO_2/pCO_2^0)^{n_{Si}} \quad (3)$$

$$F_C = F_C^0 \times (pCO_2/pCO_2^0)^{n_C} \quad (4)$$

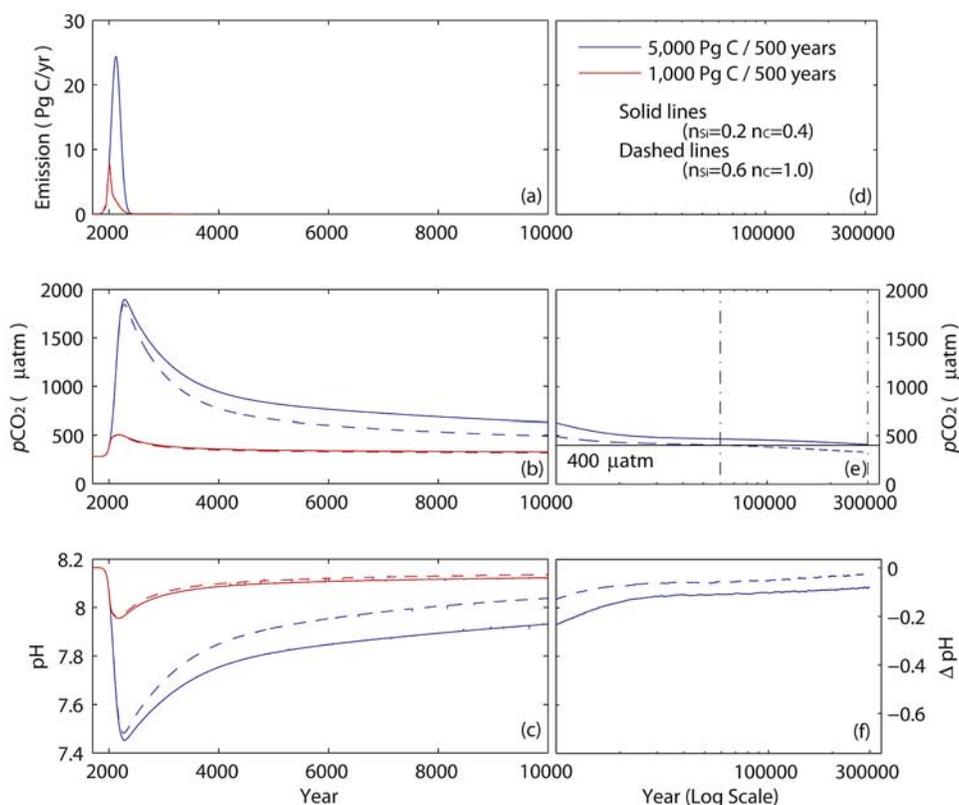
where  $F_{Si}^0$  and  $F_C^0$  (initial silicate and carbonate weathering flux) values are  $5 \times 10^{12}$  mol C/yr [Walker and Kasting, 1992] and  $12 \times 10^{12}$  mol C/yr [Morse and Mackenzie, 1990]. The default silicate and carbonate weathering parameters ( $n_{Si}$  and  $n_C$ ) were set at 0.2 and 0.4 [Zeebe et al., 2008; Zachos et al., 2008]. For the simulations under the assumption of strong weathering feedbacks,  $n_{Si}$  and  $n_C$

were raised to 0.6 and 1.0. We assessed the effects of weathering on  $pCO_2$  and ocean pH with four sets of settings: (1) default ( $n_{Si} = 0.2$ ,  $n_C = 0.4$ ), (2) enhanced silicate weathering ( $n_{Si} = 0.6$ ,  $n_C = 0.4$ ), (3) enhanced carbonate weathering ( $n_{Si} = 0.2$ ,  $n_C = 1.0$ ) and (4) enhanced silicate and carbonate weathering ( $n_{Si} = 0.6$ ,  $n_C = 1.0$ ). The parameterizations used in this study cover the range of weathering fluxes as a function of  $pCO_2$  prescribed in the GEOCARB II and GEOCARB II+FERT model [Berner, 2001] and the Walker and Kasting [1992] model (Figure 1). The abbreviation ‘FERT’ here refers to a feedback that includes a possible CO<sub>2</sub> effect on weathering by plants [Berner, 2001].

### 3. Results

[7] On the timescale of our simulations (10,000 years) the individual effect of enhanced carbonate ( $n_{Si} = 0.2$ ,  $n_C = 1.0$ ) and enhanced silicate weathering ( $n_{Si} = 0.6$ ,  $n_C = 0.4$ ) on CO<sub>2</sub> drawdown was small and barely distinguishable from each other, although carbonate weathering is slightly more effective. Thus, we will only present the results of the simulations in which silicate and carbonate weathering fluxes were simultaneously increased ( $n_{Si} = 0.6$ ,  $n_C = 1.0$ ) (Figure 2).

[8] Under the default condition, predicted maximum  $pCO_2$  ( $pCO_2^{Max}$ ) and maximum surface ocean pH decline ( $\Delta pH$ ) range from  $\sim 400$  to  $\sim 2,300$   $\mu atm$  and 0.14 to 0.78 units (Figures 3a and 3b). The magnitude of both  $pCO_2^{Max}$  and  $\Delta pH$  is dependent on the amount of total C emissions as well as the release time. In contrast, predicted atmo-



**Figure 2.** Simulated response of anthropogenic C emissions. Color coding by blue and red, respectively, indicates the model results under the emission scenario of 5,000 Pg C and 1,000 Pg C released over 500 years. (a and d) C emission scenarios and response of (b and e) atmospheric  $p\text{CO}_2$  ( $\mu\text{atm}$ ) and (c and f) surface ocean pH and pH decline from a pre-industrial value of 8.16 (denoted as  $\Delta\text{pH}$ ). Note that, in Figures 2e and 2f, the time axis (from year 10,000 to 300,000) is logarithmic and only the results under the emission scenario of 5,000 Pg C released over 500 years are shown. Solid and dashed trajectories represent simulations under default ( $n_{\text{Si}} = 0.2$  and  $n_{\text{C}} = 0.4$ ) and enhanced ( $n_{\text{Si}} = 0.6$  and  $n_{\text{C}} = 1.0$ ) weathering feedback. Two vertical dash-dot-lines in Figure 2e indicate the timing when  $p\text{CO}_2$  simulated under default and enhanced weathering parameterizations returns to  $400 \mu\text{atm}$  (roughly year 300,000 and 60,000).

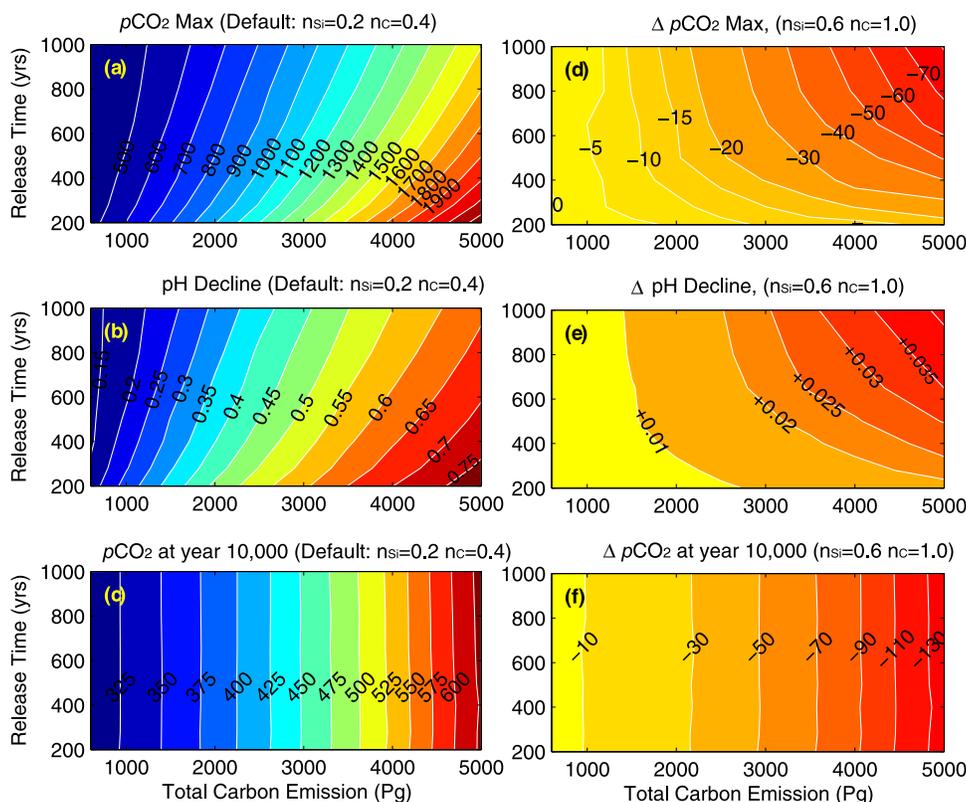
spheric  $\text{CO}_2$  in year 10,000 ( $p\text{CO}_2^{10\text{K}}$ ), ranging from  $\sim 300$  to  $\sim 630 \mu\text{atm}$ , showed dependency only on the amount of total C emissions (Figure 3c). The reason for this is that the total C input affects the quasi-steady state established in the carbon cycle after 10,000 year, while the release time does not (if much shorter than 10,000 years).

[9] Enhanced weathering generally has a minor effect on  $\text{CO}_2$  drawdown (Figure 3d). Reductions in  $p\text{CO}_2^{\text{Max}}$  are less than  $40 \mu\text{atm}$  unless large amounts of C ( $> 3,500 \text{ Pg C}$ ) are released over relatively long periods of time ( $> 500$  years). The maximum drawdown of atmospheric  $\text{CO}_2$  before it equilibrates with the ocean by enhanced weathering is only of the order of 5% (under a scenario of 5,000 Pg C total emissions released over 1,000 years). Concomitantly ocean acidification was mitigated by 0.04 pH units under this emission scenario. This reduction is, however, negligible relative to  $\sim 0.65$  units of pH decline under the default weathering fluxes (Figures 3b–3e).

[10] The effect of enhanced weathering is comparatively more important on timescales of thousands of years. The trajectories shown in Figure 2 indicate that the effectiveness of enhanced  $\text{CO}_2$  drawdown increases after year  $\sim 2,300$ . The magnitude of  $\text{CO}_2$  drawdown by weathering is enhanced under large C emissions, because weathering fluxes increase

at higher atmospheric  $\text{CO}_2$  levels (Figure 1). Enhanced weathering under the emission of 5,000 Pg C results in a decrease of  $p\text{CO}_2^{10\text{K}}$  values by as much as  $140 \mu\text{atm}$ , representing a  $\sim 22\%$  reduction relative to  $p\text{CO}_2^{10\text{K}}$  values under the default condition (Figures 3c–3f). The return of seawater pH toward pre-perturbation state is also facilitated by enhanced weathering on millennial timescales. Seawater pH under the default condition rebounds to 7.93 units in year 10,000 (Figure 2c). When a strong weathering feedback is assumed, however, the same pH is reached in year 5,250. The range of weathering parameterizations yields about 0.1 units of difference in ocean pH after year  $\sim 3,000$ .

[11] We also conducted 300,000-year long simulations to characterize the uncertainties due to weathering parameterizations in determining ‘restoring time’, the time when  $p\text{CO}_2$  will subside to climatically relevant levels after the anthropogenic perturbation (Figures 2d–2f). Under total emissions of 5,000 Pg C released over 500 years with default weathering fluxes, it takes  $\sim 300,000$  years for  $p\text{CO}_2$  to return to  $400 \mu\text{atm}$ . However, under the assumption of a strong weathering feedback, the  $400 \mu\text{atm}$  level is reestablished in  $\sim 60,000$  years. Thus the range of weathering parameters used in this study (Figure 1) can produce



**Figure 3.** (left) Expected (a) maximum  $p\text{CO}_2$  ( $\mu\text{atm}$ ), (b) maximum pH decline of average surface seawater and (c)  $p\text{CO}_2$  in year 10,000 as a function of total C emissions (Pg C) and release time (years) under default weathering parameterization ( $n_{\text{Si}} = 0.2$  and  $n_{\text{C}} = 0.4$ ). (right) Differences in (d) maximum  $p\text{CO}_2$ , (e) maximum pH decline and (f)  $p\text{CO}_2$  in year 10,000 between model runs under the default and enhanced weathering parameterizations ( $n_{\text{Si}} = 0.6$  and  $n_{\text{C}} = 1.0$ ).

uncertainties up to  $\sim 240,000$  years in the restoring time under the same emission scenario.

#### 4. Discussions and Conclusions

[12] Our simulations suggest that  $\text{CO}_2$  drawdown by enhanced weathering becomes significant only on millennial timescales in response to large C emissions. But on centennial timescales, the  $\text{CO}_2$  drawdown by weathering is small regardless of the magnitude of C emissions. Because  $\text{CO}_2$  drawdown by weathering proceeds on millennial or longer timescales, this negative feedback mechanism can only remove negligible amounts of atmospheric  $\text{CO}_2$  before the equilibration between the atmosphere and the ocean, which occurs within a few centuries [Archer *et al.*, 1998; Archer, 2005]. Thus, continental weathering has little control on the immediate  $p\text{CO}_2$  rise and ocean acidification due to future C emissions. Under the assumption of a weak weathering feedback, we predict average surface ocean pH to be lowered anywhere from 0.64 to 0.78 units for total emissions of 5,000 Pg C, depending on the release time. Under the assumption of a strong weathering feedback, the corresponding decline is estimated to be 0.60 to 0.77 units (Figures 3b–3e).

[13] The effectiveness of continental weathering on  $\text{CO}_2$  drawdown over centennial to millennial timescales demonstrated by our simulations agrees with previous studies by Lenton and Britton [2006]. Their simulations under  $\sim 4,000$  Gt C of emissions show that effect of carbonate and

silicate weathering on  $\text{CO}_2$  drawdown, although small (up to  $\sim 3\%$  and  $\sim 9\%$  enhancement for carbonate and silicate weathering), is best highlighted on timescales of  $10^3$ – $10^4$  and  $10^5$ – $10^6$  years, respectively. Dual enhancement of carbonate and silicate weathering further improved  $\text{CO}_2$  sequestration capability (by  $\sim 3\%$  at most) on  $10^3$ – $10^4$  year timescales. Archer *et al.* [2005] also suggests that carbonate weathering is important on timescales of thousands of years, whereas effect of silicate weathering is dominant on timescales over hundreds of thousands of years.

[14] The selection of weathering feedback strength is expected to be important for the simulations over long timescales. Our long-term simulations (Figures 2d–2f) show huge uncertainties caused by weathering parameters in determining the restoring time. The most important variables controlling the restoring time are, for example, the total anthropogenic C input and variations in Earth's orbital parameters. But, for a given emission scenario, the weathering parameters are critical for the predicted restoring time. A recent study by Lunt *et al.* [2008] suggests that below a threshold  $p\text{CO}_2$  value of  $400 \mu\text{atm}$ , Northern Hemisphere glaciations can be successfully triggered. In our simulations, the different weathering parameters lead to uncertainties of up to  $\sim 240,000$  years in the predicted restoring time for  $p\text{CO}_2$  to drop below  $400 \mu\text{atm}$  (Figure 2e).

[15] Archer and Ganopolski [2005] predicted that the onset of the next glacial period can be suppressed for the upcoming 500,000 years if 5,000 Gt of C are released to the atmosphere, based on an atmosphere-ocean-vegetation model

coupled to an ice sheet model. We believe that their overall timescale (~500,000 years) is probably more realistic than ours (60,000 to 300,000 years) because their results are based on the combined influence of elevated  $p\text{CO}_2$  as well as low insolation variability due to weak orbital forcing expected in the future [Loutre and Berger, 2000]. In contrast, our model does not include orbital forcing and is less complex. Nonetheless, our results show that weathering parameters can result in significant uncertainties in long-term predictions of atmospheric  $\text{CO}_2$  levels in response to massive anthropogenic C emissions.

[16] In summary, continental weathering has a minor effect on future  $p\text{CO}_2$  and ocean acidification on timescales of centuries, regardless of the assumed weathering feedback strength. This implies that uncertainties in predicting immediate atmosphere-ocean response to future C emissions due to continental weathering are small. Weathering processes are too slow to effectively alleviate short-term consequences of anthropogenic C emissions. Avoiding such consequences requires establishment of emission targets via policy protocols. On longer timescales, different parameterizations introduce huge uncertainties (up to hundreds of thousands of years) in the predicted timing when  $p\text{CO}_2$  will return to climatically relevant threshold values. In order to reduce these uncertainties, a better understanding of weathering processes is required to improve parameterizations.

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## References

- Archer, D. (2005), Fate of fossil fuel  $\text{CO}_2$  in geologic time, *J. Geophys. Res.*, *110*, C09S05, doi:10.1029/2004JC002625.
- Archer, D., and A. Ganopolski (2005), A movable trigger: Fossil fuel  $\text{CO}_2$  and the onset of the next glaciation, *Geochem. Geophys. Geosyst.*, *6*, Q05003, doi:10.1029/2004GC000891.
- Archer, D., H. Khesghi, and E. Maier-Reimer (1998), Dynamics of fossil fuel  $\text{CO}_2$  neutralization by marine  $\text{CaCO}_3$ , *Global Biogeochem. Cycles*, *12*, 259–276.
- Berner, R. A. (2001), Geocarb III: A revised model of atmospheric  $\text{CO}_2$  over Phanerozoic time, *Am. J. Sci.*, *301*, 182–204.
- Berner, R. A., A. C. Lasaga, and R. M. Garrels (1983), The carbonate-silicate geochemical cycle and its effect on atmospheric carbon dioxide over the past 100 million years, *Am. J. Sci.*, *283*, 641–683.
- Caldeira, K., and M. E. Wickett (2003), Oceanography: Anthropogenic carbon and ocean pH, *Nature*, *425*, 365.
- Caldeira, K., and M. E. Wickett (2005), Ocean model predictions of chemistry changes from carbon dioxide emissions to the atmosphere and ocean, *J. Geophys. Res.*, *110*, C09S04, doi:10.1029/2004JC002671.
- Hoegh-Guldberg, O., et al. (2007), Coral reefs under rapid climate change and ocean acidification, *Science*, *318*, 1737–1742.
- Intergovernmental Panel on Climate Change (IPCC) (2007), *Climate Change 2007: Synthesis Report. Contribution of Working Groups I, II and III to the Fourth Assessment Report of the Intergovernmental Panel on Climate Change*, edited by Core Writing Team, R. K. Pachauri, and A. Reisinger, 104 pp., Intergov. Panel on Clim. Change, Geneva, Switzerland.
- Lenton, T. M., and C. Britton (2006), Enhanced carbonate and silicate weathering accelerates recovery from fossil fuel  $\text{CO}_2$  perturbations, *Global Biogeochem. Cycles*, *20*, GB3009, doi:10.1029/2005GB002678.
- Loutre, M., and A. Berger (2000), Future climatic changes: Are we entering an exceptionally long interglacial?, *Clim. Change*, *46*, 61–90.
- Lunt, D. J., G. L. Foster, A. M. Haywood, and E. J. Stone (2008), Late Pliocene Greenland glaciations controlled by a decline in atmospheric  $\text{CO}_2$  levels, *Nature*, *454*, 1102–1105.
- Marland, G., T. A. Boden, and R. J. Andres (2007), Global, regional, and national fossil-fuel  $\text{CO}_2$  emissions, <http://cdiac.ornl.gov/trends/emis/overview.html>, Carbon Dioxide Inf. Anal. Cent. Oak Ridge Natl. Lab., Oak Ridge, Tenn.
- Montenegro, A., V. Brovkin, M. Eby, D. Archer, and A. J. Weaver (2007), Long term fate of anthropogenic carbon, *Geophys. Res. Lett.*, *34*, L19707, doi:10.1029/2007GL030905.
- Morse, J. W., and F. T. Mackenzie (1990), *Geochemistry of Sedimentary Carbonates*, *Dev. Sedimentol.*, vol. 48., Elsevier, New York.
- Orr, J. C., et al. (2005), Anthropogenic ocean acidification over the twenty-first century and its impact on calcifying organisms, *Nature*, *437*, 681–686.
- Ridgwell, A. J., and U. Edwards (2007), Geological carbon sinks, in *Greenhouse Gas Sinks*, edited by D. S. Reay et al., pp. 74–97, CAB Int., Cambridge, Mass.
- Riebesell, U., I. Zondervan, B. Rost, P. D. Tortell, R. E. Zeebe, and F. M. M. Morel (2000), Reduced calcification of marine plankton in response to increased atmospheric  $\text{CO}_2$ , *Nature*, *407*, 364–367.
- Rogner, H.-H. (1997), An assessment of world hydrocarbon resources, *Annu. Rev. Energy Environ.*, *22*, 217–262.
- Sundquist, E. T. (1986), Geologic analogs: Their value and limitations in carbon dioxide research, in *The Changing Carbon Cycle: A Global Analysis*, edited by J. R. Trabalka and D. E. Reichle pp. 371–402, Springer, New York.
- Tyrrell, T., J. G. Shepherd, and S. Castle (2007), The long-term legacy of fossil fuels, *Tellus, Ser. B*, *59*, 664–672.
- Walker, J. C., and J. F. Kasting (1992), Effects of fuel and forest conservation on future levels of atmospheric carbon dioxide, *Palaeogeogr. Palaeoclimatol. Palaeoecol.*, *97*, 151–189.
- Walker, J. C. G., P. B. Hays, and J. F. Kasting (1981), A negative feedback mechanism for the long-term stabilization of Earth's surface temperature, *J. Geophys. Res.*, *86*, 9776–9782.
- White, A. F., and A. E. Blum (1995), Effects of climate on chemical weathering in watersheds, *Geochim. Cosmochim. Acta*, *59*, 1729–1747.
- Zachos, J. C., G. R. Dickens, and R. E. Zeebe (2008), An early Cenozoic perspective on greenhouse warming and carbon-cycle dynamics, *Nature*, *451*, 279–283.
- Zeebe, R. E., and J. C. Zachos (2007), Reversed deep-sea carbonate ion basin gradient during Paleocene-Eocene thermal maximum, *Paleoceanography*, *22*, PA3201, doi:10.1029/2006PA001395.
- Zeebe, R. E., J. C. Zachos, K. Caldeira, and T. Tyrrell (2008), Carbon emission and acidification, *Science*, *321*, 51–52.

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