#### **OS11B** MCC: Hall D Monday 0830h

Impacts of Air/Sea Exchange on **Biogeochemical Processes in the** Ocean II Posters (joint with A, B, GC)

Presiding: W M Landing, Florida State University; C Measures, University of Hawaii; G A Cutter, Old Dominion University

OS11B-0222 0830h POSTER

### Heat fluxes, overturning, and the uptake of anthropogenic carbon in the North Atlantic

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Duscraption of the overturning circulation on the anthropogenic carbon sink in the North Atlantic is in-vestigated with a simple box model. The net air-sea flux of anthropogenic carbon is the result of two op-posing fluxes: one is uptake caused by the disequi-librium between the rapidly rising atmospheric pCO<sub>2</sub> and the dissolved carbon content in the ocean, depend-ing mainly on the water exchange rate between mixed layer and interior North Atlantic basin, that is directed out of heat within the Atlantic basin, that is directed out of the ocean, contrary to conventional wisdom. It is re-lated to cooling and freshening of surface water on its way north, causing a latitudinal gradient in the ratio of seawater alkalinity to total dissolved inorganic car-bon. This flux depends strongly on the vertical struc-ture of the upper branch of the overturning circulation and on the distribution of under- and super-saturation of CO<sub>2</sub> in Atlantic surface waters. A data-based esti-mate of anthropogenic carbon inventory in the North Atlantic is consistent with a dominance of the disequi-librium flux over the heat-flux-related out gassing at the present time, but in our model does not place a strong Atlantic is consistent with a dominance of the disequi-librium flux over the heat-flux-related outgassing at the present time, but, in our model, does not place a strong constraint on the net anthropogenic air-sea flux. Sta-bilization of the atmospheric  $pCO_2$  on a higher level will change the relative role of the two opposing fluxes, making the North Atlantic a source of anthropogenic rechergence the attmospheric structure. carbon to the atmosphere.

## OS11B-0223 0830h POSTER

#### Dry Deposition of Trace Elements to the Western North Atlantic

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States The concentrations of selected elements, including AI, Fe, Na, Sb, Se V, and Zn, were determined for bulk aerosol samples collected at Bermuda between 1988 and 1993 as part of the Atmospheric/Ocean Chemistry Ex-periment (AEROCE). These data were combined with parameterized dry deposition velocities to investigate trace element recycling on atmospheric sea salt. Dur-ing July and August, which are in the high dust season, > 10% of the monthly mean aerosol Na can be ascribed to crustal sources rather than sea salt, but in general the contribution of mineral dust to Na dry deposition is small ( $\sim 1\%$ ) relative to that from sea salt. In fall and winter,  $\sim 10$  to 20% of the aerosol Al, a commonly used indicator of mineral dust, is attributable to recy-cled sea salt, but the dust fluxes then are small, and longer-term averages show a recycling effect of only 1 to 5% on dry deposition. The percentage of recycled Fe appears to be even less than for Al, reaching a max-imum of  $\sim 5\%$  per month from December to February and averaging only a few percent per day when calcu-lated over the duration of the study. Non-crustal/non-sea salt (NC/NSS) sources account for > 70% of the aerosol Sb, Se, V, and Zn, but differences in the dry deposition velocities for three particle types, which do still need to be verified, strongly affect their air/sea exchange rates. According to the simple model calcu-lations, dust can account for more than half of the new dry deposition inputs of V and Zn, but the amounts of Sb and Se recycled on sea spray approach or exceed their new inputs from dust and the NSS/NC sources. The concentrations of selected elements, including

### OS11B-0224 0830h POSTER

#### The Concentration of Hafnium in Seawater: a Comparison Between the Arctic Ocean and the Northwestern Pacific

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<sup>2</sup>Department of Earth Sciences, University of Oxford, Parks Road, Oxford OX1 3PR, United Kingdom The concentrations and isotopic composition of hafnium in geological samples and ocean sediments have been studied extensively. However, the sources of Hf and its behaviour in the aquatic environment are poorly understood. We established a low blank, high efficiency technique to determine the Hf concentration in seawater. Isotope dilution measurements by multiple collector inductively coupled plasma mass spectrometry (MC-ICPMS) offer a method to obtain accurate concen-tration measurements from sample volumes of 2 litres. We will present water column profiles of Hf con-centration from two oceanic realms with different hy-drographic features and particle concentration (samples from Sea-Bird SBE 9 2000, Swedish AO-01 expedition, 2002 IOC western Pacific cruise). In the Arctic Ocean with its perennial sea ice cover the suspended particle cocean basins, which is a result of the low dust input and the low productivity. The renewal rate of deep water is slow (isolation ages of 250 to 450 years; Schlosser et al. 1994). In contrast, the Pacific Ocean is exposed to dust input from the Asian deserts and biological pro-tuctivity at the margin is high. The circulation pat-terns are controlled by geostrophic forcing and inflow-ing deep waters from the Southern Ocea. The preliminary results of samples from the Arc-tic Ocean show a slight decrease with depth, in con-trast to earlier concentration measurements from the Atlantic and Pacific Ocean where Hf concentrations in-rease with depth (Godfrey et al. 1996, Mc Kelvey and Orians, 1998). The range of our measured concentra-tions (0.4 to 0.8 pmol/kg) is comparable to earlier re-sults. The higher concentration (0.7 to 0.8 pmol/kg) in the surface waters of the Arctic Ocean are proba-bly a result of reduced scavenging due to low particle concentrations. The Hf concentration data and eventu-ally the Hf isotopic compositon of sexawater will lead to a better understanding

concentrations. The Hf concentration data and eventu-ally the Hf isotopic composition of seawater will lead to a better understanding of its suitability as a tracer for past and present ocean circulation and continental weathering. References Godfrey et al., (1996), Geochimica et Cosmochimica Acta, 60, 3995-4006. McKelvey and Orians, K.J., (1998), Marine Chem-istry, 60, 245-255.

istry, 60, 245-255. Schlosser et al., (1994), Radiocarbon, 36, 327-343.

#### OS11B-0225 0830h POSTER

Measurements of Atmospheric Trace Gases Over NW Pacific During IOC 2002 Cruise - Pollutants Transported From East Asia and Biogenic Species **Emitted From Ocean** 

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The R/V Melville cruised from Osaka (Japan) on May 1st to Hawaii on Jun 5th, as a project of Inter-governmental Oceanographic Commission (IOC) 2002. During this cruise, atmospheric trace gases (O<sub>3</sub>, CO, DMS, many hydrocarbons, and halocarbons) were mea-sured. It is expected that polluted air as well as Kosa sured. It is expected that polluted air as well as Kosa from east Asia is transported to the Pacific in this sea-son. On the west Pacific near from Japan, the concen-trations of anthropogenic species were strongly affected by the meteorological condition. And the hydrocarbon concentrations have not very large difference at the east of Japan (about 145° E) and at 170° E in the same lat-itude.

concentrations have not very large different at the ease lat-itude. On 170°E, the ship cruised straight from 50°N to about 20°N, and latitudinal profile of atmospheric species in the Pacific was observed. There are clear concentration decrease from higher latitude to lower latitude for CO, O<sub>3</sub>, hydrocarbons, and short lived halocarbons. There is a large gap at around 30°N, since maritime air prevail at lower latitude. The ra-tios of hydrocarbons and CO give some information about their sources. CO and ethane have similar life-time in the atmosphere, but the ratio ([ethane]/[CO]) decreased drastically as the distance from the land. CO is supplied by the oxidation of CH<sub>4</sub> and hydrocarbons also on the ccean, in addition to the transportation of anthropogenically generated CO from land. Therefore concentration decrease of CO is smaller than that of hydrocarbons far from the anthropogenic source area. Biogenically generated species emitted from the locean (DMS, ethene, propene, isoprene, CH<sub>2</sub>Br) were atotally different from anthropogenic hydrocarbons. Higher concentrations of bogenic generated species were observed at lower latitude, where more activated biogenic activity is expected. Similar trend of ethene, propene, and DMS were observed. Especially, ethene and propene have good correlation. Since these species have relatively high reactivity in the atmosphere, they affect the local chemistry of the atmosphere on the ocean.

### OS11B-0226 0830h POSTER

### Phytoplankton Community Composition, Growth and Mortality

Rates in the North Pacific During May 2002

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of Hawaii, 1000 Pope Road, Honolulu, HI 96822, United States The IOC North Pacific expedition of May 2002 from Osaka, Japan to Honolulu, Hawaii, surveyed the region from roughly 22 - 50°N 147°E - 158°W. This survey in-cluded nine stations, at which we collected samples for analysis with microscopy and flow cytometry to char-acterize the abundance and biomass of phytoplankton and microheterotroph populations. We also conducted deckboard-incubated dilution experiments to determine the growth and mortality rates of the phytoplank-ton community at eight of the nine stations occupied. Preliminary results suggest that the stations studied can be distinguished based on phytoplankton commu-nity composition and growth response to macronutri-ent additions during incubations. Station 1, located in the Kuroshio Current (34°22.76N 147°11.84E), was characterized by a centric diatom bloom (~20°C, sur-face chlorophyll 0.7  $\mu$ g l<sup>-1</sup>) and phytoplankton growth

rates were significantly enhanced after the addition of ammonium and phosphate. Phytoplankton com-munities at four oligotrophic stations (<0.5  $\mu$ M ni-trate, <0.1  $\mu$ M phosphate) also exhibited enhanced growth rates after macronutient additions (Station 6: 30°N 170°E; Station 7: 24°N 170°E; Station 8: 26°N 175°W; and Station 9: 22°45N 158°W). Sur-face chlorophyll 1<sup>-1</sup> with surface temperatures rang-ing from 22-27°C. The phytoplankton communities at these oligotrophic stations were dominated by pi-coplankton (*Prochloreoccus* and *Synechococcus*) and small (<5  $\mu$ m) autotrophs. Finally, no nutrient limitation was observed at three of the stations (Station 2: 44°N 170°E). Surface chlorophyll at these stations aranged from 0.3 to 0.7  $\mu$ g 1<sup>-1</sup> and surface temperatures ranged from 0.3 to 0.7  $\mu$ g 1<sup>-1</sup> and surface temperatures ranged from 4.13°C. The phytoplankton at these stations were dominated by autotrophs measuring 2-10  $\mu$ m, including haptophytes and pennate diatoms. In addition, 10-20  $\mu$ m autotrophic dinoflagellates were common at Station 2. Phytoplankton growth and mortality rates at each of these stations will be discussed.

#### OS11B-0227 0830h POSTER

#### Mercury cycling over the North Pacific Ocean: Influence of South East Asia dust depositon.

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Mercury in the marine environment was investi-Mercury in the marine environment was investi-gated as part of a comprehensive study of its bio-geochemical cycle in the North Pacific Ocean on the 2002 IOC cruise. The impact of dust deposition on the concentration of mercury in surface water was as-sessed using measurements in the water column (total and gaseous dissolved species) as well as in the ma-rine boundary layer. Incubation experiments were per-formed in order to investigate the role of photochem-istry in the air-sea exchange of mercury. In addition we will present a comparative study of mercury cycling between the Pacific Ocean and the Atlantic Ocean.

## OS11B-0228 0830h POSTER

#### **Observations of Near-Surface Thermal** Structure and Turbulence Parameters

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ton 6003, New Zealand Turbulent transport in the upper few meters of the ocean is one of the major factors controlling air-sea heat, momentum, and mass exchange. A number of issues related to this exchange are poorly under-stood. For example, in low wind speed regimes, near-surface stratification can significantly affect turbulent processes. However, under more energetic conditions, processes affecting near-surface layer turbulence and thus air-sea exchange are not well quantified. Here we present observations in the near-surface layer from an experiment in Wellington Harbor, New Zealand. Continuous measurements were made from local time 1200-1700, during which time the wind dropped from 14 to 0 ms<sup>-1</sup>. During this period the air was 4 °C colder than the water. The primary results were derived using the cycling autonomous profiler SkinDeEP. Additional turbulence data were recorded using a SCAMP temperature gradient microstructure profiler and a VECTOR acoustic velocimeter. A mi-crowave radar quantified the white-capping percent-age. The data showed: (i) the turbulent albeit inho-mogeneous nature of the near-surface fluid and (ii) the change in temperature structure as the wind forcing eases. A comparison is also made with neutral condi-tions where the air and sea were at comparable temper-atures. atures

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## OS11B-0229 0830h POSTER

#### Silver in the North Pacific

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Street, Santa Cruz, CA 95064, United States Water data gathered in the North Pacific (May 2002) during the fourth International Oceanographic Commission cruise (IOC IV) are offering new insights into the biogeochemical cycling of Ag in the worlds oceans. Ag concentrations in surface waters and from nine vertical profiles will greatly supplement the lim-ited data set of Ag measurements in the Pacific Ocean, which at present consist of one site in the South Pa-cific [1], two sites in the eastern North Pacific [2], and three sites near Japan [3]. Natural biogeochemical pro-cesses govern the distribution of Ag, as is clear from its nutrient-type distribution in oceanic vertical pro-files, with depletion in surface waters and enrichment at depth. Ag has previously been shown to positively correlate with Si [2-6], and that trend will again be examined in these samples. Any deviations from this trend can be used as an indicator of anthropogenic Ag contamination. contamination.

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## OS11B-0230 0830h POSTER

#### The Atmospheric Dust Concentration and the Fractional Solubility of Atmospheric Al in seawater

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<sup>2</sup>CSIRO Land and Water , GPO Box 1666, Canberra, ACT 2601, Australia In order to determine the atmospheric dust con-centration and the fractional solubility of atmospheric aluminum (Al), aerosol samples were collected weekly on the windward side of Oahu, Hawaii between Febru-ary and June 2002. Sampling was conducted using an automatic sector controlled aerosol sampling system to prevent pumping during times when winds would bring dust from nearby islands and local sources to the sampler. Aerosols were collected on 10 replicate acid cleaned Millipore 47 mm Polycarbonate filters with a pore size 0.4  $\mu$ m at a mean pumping rate of 80 L/min. Microwave acid digestion of a complete set of filters and their aerosol load followed determination of Al by flow injection analysis (FIA) yielded a precision (1 $\sigma$ ) of 11% for the entire sampling and analysis process. Par-tial dissolution of the aerosols was determined by sus-pending a replicate filter in 60 mL of 0.2  $\mu$ m filtered was then filtered (0.2  $\mu$ m) and the amount of dissolvable Al 24 hour intervals for 4 successive days. Another repli-cate filter sampling period, the atmospheric Al concentrations and atmospheric mineral dust concen-trations observed above Hawaii were 1.2-56.1 (mg/m<sup>3</sup>) and 15-700 (mg/m<sup>3</sup>), respectively, assuming mineral

trations observed above Hawaii were 1.2-56.1  $(ng/m^3)$ and 15 700 ( ( 3) and 15-700 (ng/m<sup>3</sup>), respectively, assuming mineral aerosols are 8% Al by mass. Atmospheric dust con-centrations at Hawaii increased shortly after an intense dust outbreak was observed in Mongolia in late March. Al fractional solubility varied from 0.087-14.3% with a

mean value of 4.6%. Within the first 24 hours, Al in the aerosol samples appeared to dissolve relatively quickly (first mode). It then continued to dissolve slowly over the next three days (second mode). The total dissolvable Al (1st plus 2nd mode) as well as the total charge of dissolvable non sea salt (nss)  $K^+$ ,  $Mg^{2+}$  and  $Ca^{2+}$ was well correlated with the total charge of NO3 - and nss  $SO_4^{2-}$  (R<sup>2</sup>=0.78, R<sup>2</sup>=0.84). These results indicate that the abundance of atmospheric (NO<sub>3</sub><sup>-</sup> + nss  ${\rm SO_4}^{2-}$ ) acids is one of the major factors controlling the fractional solubility of atmospheric dust.

#### OS11B-0231 0830h POSTER

Halocarbon Production by Pelagic, Sea-Ice and Snow Communities of Micro-organisms - A Source for Halogens in the Arctic Troposphere

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<sup>9</sup>Department of Plant Ecology, Uppsala University, Villavagen 14, Uppsala SE-75236, Sweden Volatile halocarbons are trace constituents of the oceans and the atmosphere, and they contribute signif-icantly to atmospheric chemistry and the biogeochemi-cal cycling of the elements chlorine, bromine and iodine in the environment. The trace gases and their degra-dation products contribute to the atmospheric destruc-tion of ozone. An inverse correlation between bromine and ozone in the troposphere has been reported from certain areas, such as the Arctic region. Previous observations suggest that there is a major release of brominated substances into the troposphere during early spring in the Arctic Ocean. Therefore, we participated in an expedition during April and May 2002 to the Greenland Sea with I/B Oden to identify the origin of halocarbon production with emphasis on brominated compounds. Incubation experiments were performed with micro-organisms from different habitats, seawater, brine and snow. The aim was to measure the magnitude of halocarbon production and relate this to the micro-organisms in the different habitats. It has been sug-gested that the formation of organo-bromine com-pounds in snow in spring conditions may be due to the formation of reactive bromine species such as HOBF. However, tests of the brominating activity in seawater, brine and melted snow suggested no significant produc-tion of HOBr. Instead our results indicate that snow and ice communities contribute to a larger extent to the total production of halocarbons (e.g. bromoform) than previously assumed. We also found that the production total production of halocarbons (e.g. bromoform) than previously assumed. We also found that the production was higher with increasing salinity in the brine. URL: http://www.amc.chalmers.se

OS11B-0232 0830h POSTER

# Colloidal Iron, Aluminum, and DOC/DON in Surface Waters of the Northwest Pacific: Results from the 2002 NSF/IOC Cruise

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sity, Tallahassee, FL 32306 Over 70 surface seawater samples were collected on the 2002 NSF/IOC cruise between Japan and Hawaii using a towed fish peristaltic pump trace-metals clean sampling system. Samples for total dissolved Fe and AL were filtered using 0.2 um cartridge filters. Col-loidal Fe and Al were isolated using a Millipore Prep-Scale 1 kDa regenerated cellulose tangential flow ultra-filtration device. Concentration factors were 8-10. Alu-minum concentrations were measured using the lumo-gallion fluorometric technique; Fe concentrations were measured by Fe-57 isotope dilution with a Finnegan El-ement high-resolution magnetic sector ICPMS. Total dissolved Fe concentrations ranged from 0.2 to 0.6 nM and were weakly correlated with atmospheric Fe

deposition (calculated from aerosol Fe concentrations). Colloidal Fe ranged from 10-60 percent of the total dis-solved Fe and appeared to be related to atmospheric input and biological activity. We will discuss the re-lationships between the concentrations of particulate, dissolved and colloidal Fe, Al, and DOC/DON and the intertwined effects of atmospheric input, complexation by natural ligands, and physical dilution into the mixed

#### OS11B-0233 0830h POSTER

#### Atmospheric Deposition and Upper Ocean Cycling of Iron and Aluminum in the Gulf of Mexico: Results from the SWISS project

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<sup>2</sup> Louisiana Universities Marine Consortium, 8124 Highway 56, Chauvin, LA 70344, United States Three cruises have been conducted in the Gulf of Mexico for the Surface Water Iron Speciation Study (SWISS). Our goal is to investigate the processes that lead from aerosol Fe solubilization to stabilization in the water column and eventually to uptake by phyto-plankton. One cruise was conducted during the low dust, high river runoff period (spring 2001); two cruises occurred during the high dust, low runoff period (late summer 2001 and 2002). Surface seawater samples were collected using a clean towed batwing pump sys-tem and separated into particulate, total dissolved, col-loidal, and truly dissolved fractions using cartridge fil-ters and tangential flow ultrafiltration. Colloidal Fe concentrations ranged from 20-80 percent of the total soluble Fe, and were generally lower in the open Gulf waters. Aerosol samples were collected for 12-24 hour intervals onto 47 mm, 0.45 um filters using a sector-controlled system to avoid stack exhaust. The solu-bilities of aerosol Fe and Al were assessed by rinsing the filters with surface seawater or ultrapure water. The leach solutions were analyzed for trace elements and major ions. The concentrations of total aerosol Fe ranged from 5-30 nmol/m3 and correlated strongly with Al. The total soluble Fe ranged from 2-19 percent and averaged 9 percent. The soluble Fe(II) ranged from 1-60 percent of the total soluble Fe (average 17 percent). The data from the third cruise (Sep. 2002) will be dis-cussed in context with what we have learned from the first two cruises.